

Study of nanocomposites containing core-shell fillers with rigid nano-SiO₂ core in PMMA matrix

Dean Shi · Wei Yu · Robert K. Y. Li

Received: 21 August 2007 / Accepted: 30 November 2007 / Published online: 9 December 2007
© Springer Science+Business Media, LLC 2008

Through the years the subject of polymer toughening has attracted attention from both academia and industry. An effective method to toughen inherently brittle polymers is rubber toughening [1–4]. While the incorporation of rubber particles can improve the fracture toughness in brittle polymers, it will also reduce the tensile modulus and tensile strength accordingly [1]. From an application view point, one would naturally prefer to have materials that are tough and with the tensile properties maintained. The natural choice would be the incorporation of rigid fillers (such as inorganic particulates or glass fibres) and soft elastomeric inclusions simultaneously [5–7]. The reinforcement theories of fibre reinforced composites are well documented [8]. In our earlier studies [5] on short glass fibre (SGF) and rubber (an EPR rubber) filled polypropylene (PP) ternary blends, both SGF and rubber inclusion improved the impact fracture toughness of brittle PP homopolymer. However, the ability of SGF to toughen the PP matrix is well below that of the elastomer.

In recent years, studies on nanosize rigid particles modified polymer matrix composites have demonstrated that dramatic properties' improvement can be achieved with only a low level of fillers added [9–11]. Ou et al. [9] showed that with the addition of 5 wt% of silica particles (with sizes ranging from 50 to 110 nm) in nylon 6, the tensile modulus, tensile strength, elongation at break and impact strength have all been significantly improved.

Nowadays, the preparations of organic/inorganic nanoparticles with core-shell structures through both emulsion [12–15] and dispersion polymerization [16] have gained much progress. However study of the characteristics of these nanoparticles in polymer matrix as functional fillers is still rare [17–19]. In this article, we will report on the preparation of core-shell nanoparticles with rigid SiO₂ core and rubbery polybutyl acrylate (PBA) shell via in situ emulsion polymerization and its application as toughening filler of PMMA through emulsion co-precipitation method.

Silica nanoparticles with size of ~40 nm (grade D-60287) was supplied by Degussa (China) Co. Ltd. and were used after drying in vacuum at 105 °C for 12 h. The pre-treatment of silica with vinyltriethoxysilane (VETOS) was carried out using a previously published procedure [19]. The monomers, methyl methacrylate (MMA) (from Tianjin Boyou Chemical Co. Ltd.) and butyl acrylate (BA) (from Tianjin Comile Chemical Agent Co. Ltd.) were distilled under reduced pressure before use. The initiator, ammonium persulphate (APS), was recrystallized from water before use. The surfactant, sodium dodecylsulphonate (SDS); the demulsifier, AlCl₃; and the buffering agent, NaHCO₃ were used without further purification. The crosslinking agent 1,3-butanediol dimethacrylate (BDMA) was supplied by Tokyokasei (TCI) Co. Ltd. and used as received.

The emulsion polymerization was carried out in a 250 ml four-necked round-bottomed flask, which was equipped with a reflux condenser, a mechanical stirrer, a dropping funnel and a nitrogen gas inlet, respectively. In a typical synthesis, the modified nano-SiO₂ powder was dispersed into distilled water with the aid of ultrasound bath for 2 h. The initiator (APS) was dissolved in a definite amount of distilled water, and then the buffer solution of NaHCO₃ and surfactant (SDS) was added. This solution

D. Shi · R. K. Y. Li (✉)
Department of Physics and Materials Science, City University
of Hong Kong, Tat Chee Avenue, Kowloon, Hong Kong
e-mail: aprkyl@cityu.edu.hk

D. Shi · W. Yu
Faculty of Materials Science and Engineering, Hubei University,
Wuhan, China

was transferred into the reactor flask, which was charged with modified nano-SiO₂ dispersion. The initial charge was stirred at 300 rpm for 2 h before feeding with monomers. Afterward, the monomers (BA or BA/BDMA mixture) were fed into the reactor in a separate stream with constant flow rates which were slow enough to reach monomer-starved conditions. The time of feeding was 60 min and after addition of the monomers, the content was heated up to 75 °C and stirred for an additional 4 h. Then, the system was cooled to room temperature and the final latex was obtained. The latex with nano-SiO₂ particles was then mixed with a certain amount of pure PMMA latex (formulation refer to Latex 4 in Table 1) under constant stirring. Then the mixed latex was subjected to demulsification with AlCl₃ solutions, and the product was collected by suction filtration and washed thoroughly with hot water, then dried. The detailed recipes prepared in this research are given in Table 1. Basically, latexes 1, 2, 3 and 4 were first produced. Subsequently, latex 4 was mixed with latex 1, 2 and 3 to obtain blends 5, 6 and 7, respectively.

Figure 1a, b shows the TEM micrographs of Latex 2 and Latex 3, respectively. It was found that the micelles in Latex 2 without the crosslinking agent (BDMA) had relatively larger particle size than those in Latex 3 with BDMA. This was because the micelles with uncrosslinked shell are much softer than those with crosslinked ones. These uncrosslinked micelles will be deformed and flattened more easily on the copper grid when the TEM samples were prepared. And it can be clearly found in the micrograph of Latex 2 that not single but SiO₂ particle

clusters were encapsulated in the individual micelle. This result is similar to other reported works [12].

Figure 2 is the DSC heating curve of pure PMMA (obtained from Latex 4) and the PMMA blends with different kinds of fillers. It could be found that for pure PMMA, the glass transition temperature (*T_g*) is around ~114 °C. When added with pure PBA rubber (Blend 5), the *T_g* value decreased to ~103 °C because of the low *T_g* of PBA (~−76 °C) [20, 21]. While adding core-shell particle with SiO₂ core and PBA shell in PMMA matrix (Blend 6), the *T_g* of PMMA again increases to ~108 °C. The *T_g* increase is due to SiO₂ core acting as entanglement points of the PBA shell which will strongly prohibit the movement of the PBA molecular chains and increase the *T_g* of PBA. Furthermore, when the PBA shell was cross-linked with BDMA in which the PBA molecular chain will be further restricted and results in further *T_g* increased of the rubber shell, as well as the *T_g* of PMMA blends (~115 °C) (Blend 7).

The storage modulus (*E'*) versus temperature plots for pure PMMA and the different blends are compared in Fig. 3. It can be found that the storage moduli of blends with core-shell fillers (Blend 6 and Blend 7) are significantly higher than those with only pure rubber toughener system (Blend 5). When the temperature is lower than the *T_g* of PBA, the storage moduli of blends with core-shell toughener (Blend 6 and Blend 7) are even higher than that of pure PMMA. This is due to the combined reinforcement effects of the glassy rubber shell and rigid SiO₂ core of the core-shell fillers. When the temperature exceeds the *T_g* of PBA, the storage moduli of these blends decreased rapidly and finally become lower than that of pure PMMA. Comparing the two blend systems with core-shell fillers, the storage modulus of the one with crosslinked PBA shell (Blend 7) is slightly higher than that with uncrosslinked PBA shell particles (Blend 6) within the whole temperature zone.

The TGA results shown in Fig. 4 show that the thermal stability of blend with cross-linked core shell particles (Blend 7) is also higher than that with uncrosslinked ones (Blend 6). These results indicated that the crosslinking of rubber shell of core-shell particles will enhance not only its modulus but also its thermal stability.

In summary, core-shell particles with rigid SiO₂ core and rubbery PBA shell have been prepared through in situ emulsion polymerization method, and these core-shell particles were incorporated into PMMA matrix. The existence of rigid SiO₂ core in the core-shell particles can dramatically enhance the modulus of the blends compared to the addition of pure PBA rubber particles alone. The crosslinking of the rubber PBA shell of the core-shell particles does not only improve the modulus but also the thermal stability of the blends.

Table 1 Emulsion formulation of SiO₂-PBA core-shell rubber and SiO₂-PBA/PMMA blends

Latex						
Sample	BA (g)	NaHCO ₃ (mg)	APS (mg)	SDS (g)	SiO ₂ (g)	BDMA (mg)
Latex 1	3	8	10	0.1	0	0
Latex 2	2.1	8	10	0.1	0.9	0
Latex 3	2.1	8	10	0.1	0.9	4.2
Blend						
	MMA (g)	NaHCO ₃ (mg)	APS (mg)	SDS (g)	SiO ₂ (mg)	
Latex 4	3	8	10	0.1	0	
Blend						
Blend 5 ^a	Latex 1 + Latex 4					
Blend 6 ^b	Latex 2 + Latex 4					
Blend 7 ^c	Latex 3 + Latex 4					

^a The weight ratio between PBA and PMMA is 1:9; ^b The weight ratio of PBA + SiO₂ and PMMA is 1:9; ^c The weight ratio of PBA + SiO₂ and PMMA is 1:9

Fig. 1 TEM photograph of core-shell PBA/SiO₂ micelles: (a) with BDMA (Latex 2); and (b) without BDMA (Latex 3)

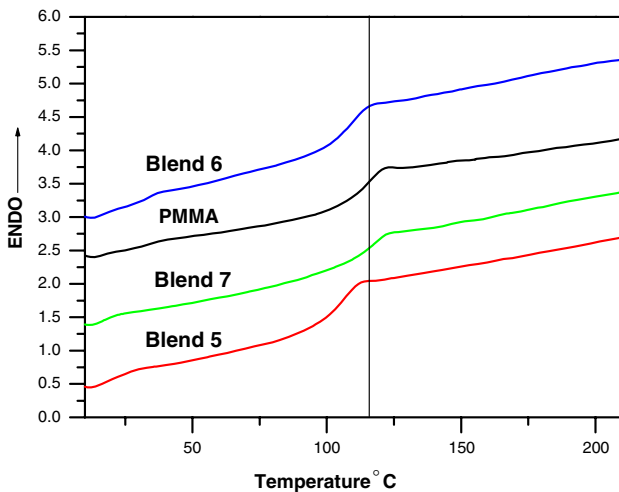
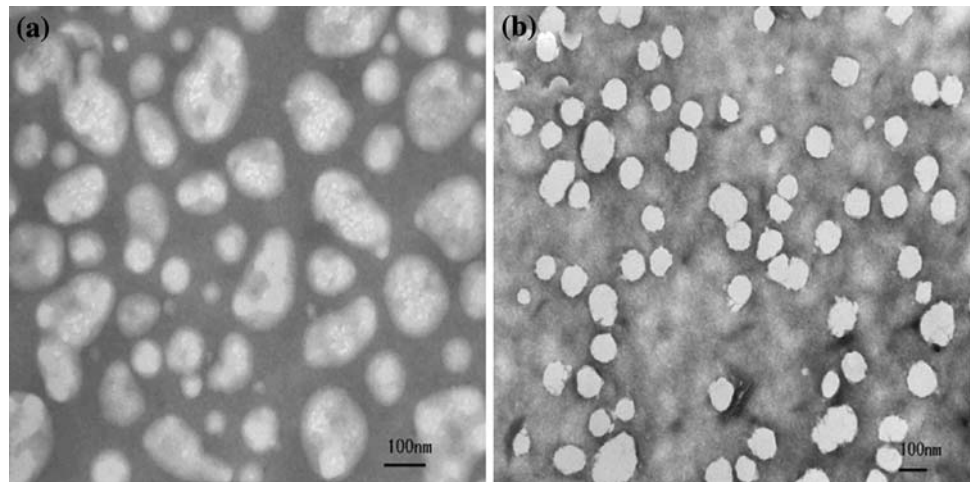


Fig. 2 DSC heating curves of PMMA and the different blends

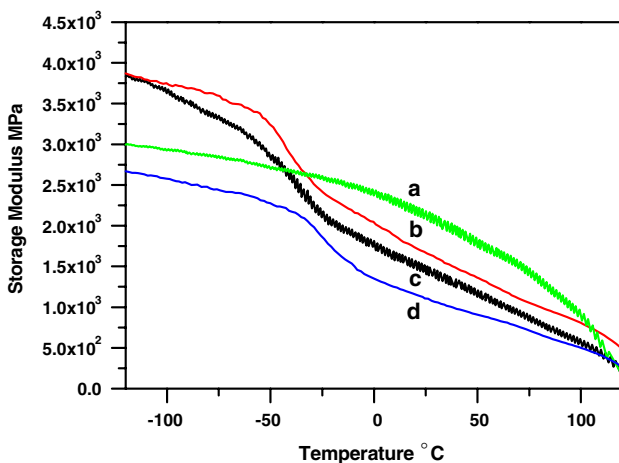


Fig. 3 Effect of temperature on the storage modulus for: (a) Pure PMMA; (b) Blend 7; (c) Blend 6; and (d) Blend 5

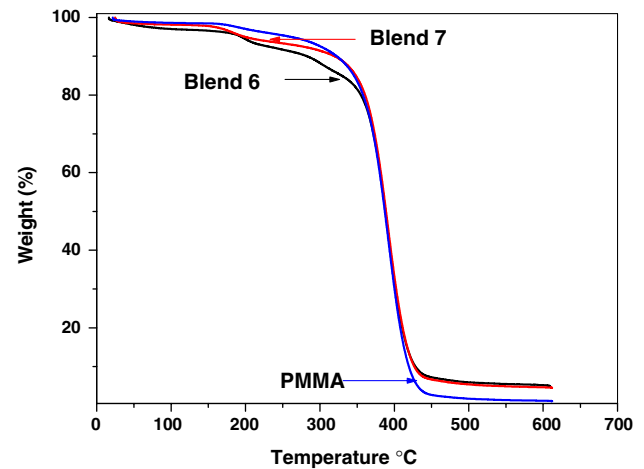


Fig. 4 TGA results of PMMA nanocomposites with core-shell particles

Acknowledgements The work described in this article is supported by the Research Grants Council of the Hong Kong Special Administrative Region, China (Project No. CityU 117205) and the National Science Foundation of China (Grant Nos. 50373011 and 50763024).

References

1. Tam WY, Cheung T, Li RKY (1996) *Polym Test* 15:363
2. Wu S (1985) *Polymer* 26:1855
3. Liu ZH, Li RKY, Tjong SC, Qi ZN, Wang FS, Choy CL (1998) *Polymer* 39:4433
4. Liu ZH, Li RKY, Tjong SC, Choy CL, Zhu XG, Qi ZN, Wang FS (1999) *Polymer* 40:2903
5. Tam WY, Cheung TYH, Li RKY (2000) *J Mater Sci* 35:1525
6. Ching ECY, Li RKY, Tjong SC, Mai YW (2003) *Polym Eng Sci* 43:558
7. Liang JZ, Li RKY, Tjong SC (2000) *Polym Eng Sci* 40:2105
8. Chou TW (1992) *Microstructural design of fiber composites*. Cambridge University Press, Cambridge
9. Ou Y, Yang F, Yu ZZ, (1998) *J Polym Sci Part B Polym Phys* 36:789

10. Ash BJ, Siegel RW, Schadler LS (2004) *Macromolecules* 37:1358
11. Zhang M, Singh RP (2004) *Mater Lett* 58:408
12. Zhou J, Zhang SW, Qiao XG, Li XQ, Wu LM (2006) *J Polym Sci Part A: Polym Chem* 44:3202
13. Wang H, Shi T, Zhai L (2006) *J Appl Polym Sci* 102:1729
14. Zeng Z, Yu J, Guo ZX (2004) *Macromol Chem Phys* 205:2197
15. Cheng XJ, Chen M, Zhou SX, Wu LM (2006) *J Polym Sci Part A: Polym Chem* 44:3807
16. Mahdavian AL, Ashjari M, Makoo AB (2007) *Eur Polym J* 43:336
17. Xie XL, Li BG, Pan ZR, Li RKY, Tjong SC (2001) *J Appl Polym Sci* 80:2105
18. Xie XL, Li RKY, Liu QX, Mai YW (2004) *Polymer* 45:2793
19. Xie XL, Tang CY, Zhou XP, Li RKY, Yu ZZ, Zhang QX, Mai YW (2004) *Chem Mater* 16:133
20. Itagaki H, Ochiai A, Nakamori T (2006) *Eur Polym J* 42:1939
21. Chen Z, Gong K (2002) *J Appl Polym Sci* 84:1499